

171.312 STATISTICAL PHYSICS AND THERMODYNAMICS
FINAL EXAM
Monday December 12, 2011

[18 points]

1) Provide brief answers to all of the following questions:

- (a) What is the “Fundamental Assumption”?
- (b) What is the 0th law of thermodynamics?
- (c) What is the 1st law of thermodynamics?
- (d) What is the 2nd law of thermodynamics?
- (e) What is the 3rd law of thermodynamics?
- (f) Write an expression for the “quantum concentration” and state its relevance for ideal and degenerate gases.
- (g) What choice of reservoir material maximizes the Carnot efficiency of a heat engine?
- (h) For a three dimensional solid with N atoms, what is the maximum number of phonon modes?
- (i) For a cube of volume $V = L^3$, what is the maximum number of electromagnetic modes?

[10 points]

2) The following question involves tosses of a fair coin with a 50% chance of heads and a 50% chance of tails.

- (a) If a coin is flipped three times, what is the probability that heads will come up 2/3 of the coin tosses and tails 1/3 of the coin flips?
- (b) For 6 flips, what is the probability that the coin will come up 2/3 heads and 1/3 tails?
- (c) For $N=10^{22}$ flips what is the probability of 2/3 heads and 1/3 tails? [Please provide only a very rough approximation here! No detailed calculations, please!]
- (d) What is the implication of (a) through (c) for equilibrium and the determination of the multiplicity of large thermodynamic systems?

[5 points]

3) A system at temperature T includes two particular states, **A** and **B**. The energy of state **A** is ε_A and the energy of state **B** is $\varepsilon_B = \varepsilon_A + \delta\varepsilon$. What is the relative probability of the system being in state **A** relative to being in state **B**?

[10 points]

4) Consider a system of $N \gg 1$ non-interacting particles in which the energy of each particle can have two (and only two) distinct values: 0 and E ($E > 0$). The occupation numbers of the zero and E energy levels are n_0 and n_1 respectively. The fixed total energy of the system is U .

- (a) Find the entropy of the system.
- (b) Find the temperature as a function of U .
- (c) For what range of values of n_0 is $\tau < 0$?

[7 points]

5) Two separate containers, P and Q, contain different kinds of degenerate gases. Both of the different types of degenerate gases are in their ground state. The kinetic energy of the degenerate gas in P is much higher than that of Q.

- What kind of degenerate gas is P and what kind is Q? Explain.
- Which will have the higher thermal conductivity?
- One of the two gases dramatically changes its behavior below 2.17 K. Name this gas.

[10 points]

6) The entropy of a substance is σ_l when in its liquid phase and σ_g when in its gas phase, both for temperature τ and constant pressure, p .

- What is the heat of vaporization of the substance?
- If the enthalpy of the gas is H_g , provide an expression for the enthalpy of the liquid, H_l .
- What is the Gibbs function of the liquid, G_l ?
- What is the Gibbs function of the gas, G_g ?
- Prove that the Gibbs function does not change in a reversible isothermal and isobaric process.

[10 points]

7) On a single graph, sketch the distribution function $f(\varepsilon)$ for a degenerate Bose gas, a degenerate Fermi gas, and a classical gas, all as a function of $(\varepsilon - \mu)/\tau$ (which can be positive or negative). Assume that the temperature is somewhat above absolute zero. Label all important features.

[15 points]

8) Consider a gas of N noninteracting, spin $\frac{1}{2}$ fermions of mass M , initially in a box of volume V_i at temperature $\tau_i = 0$. Let the gas expand irreversibly into a vacuum, without doing work, to a final (thermally isolated) volume V_f .

- What is the average energy per particle before the expansion $\langle \varepsilon_i \rangle / N$ as a function of the initial density N / V_i ?
- What is the average energy per particle after the expansion $\langle \varepsilon_f \rangle / N$ as a function of the final temperature τ_f , assuming that the gas expanded enough to be an ideal gas after the expansion?
- Is energy conserved? I.e., does $\langle \varepsilon_i \rangle = \langle \varepsilon_f \rangle$?
- What is the final temperature, τ_f , of the gas after expansion as a function of mass M and initial density N / V_i ?
- What range of final temperature of the gas τ_f after expansion, as a function of mass M and final density N / V_f , is consistent with the assumption that the gas expanded enough to be treated as an ideal gas after the expansion?
- What is the minimum final volume, V_f ?

[15 points]

9) A simple theory of the thermodynamics of a ferromagnet expresses the free energy, F , as a function of the magnetization, M in the following form:

$$F = -HM + F_0 + a(T - T_c)M^2 + bM^4$$

where H is the magnetic field, and F_0 , a and b are all positive constants, T is temperature, and T_c is the critical temperature.

- (a) Explain what condition on the free energy F determines the thermodynamically most probable value for the magnetization M in equilibrium?
- (b) Determine the equilibrium value for M assuming that $T - T_c \gg 2bM^2 / a > 0$
- (c) Sketch a graph of M versus T from (b)
- (d) Instead of the assumption in part (b), determine the equilibrium value for M for the assumption that $H = 0$. Explicitly check that your equilibrium solution(s) is/are stable.